

Presentation for Topical Panel

**Nano and Biological Technology and Their
Potential Applications**

December 3, 2008



Presented by:

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Report Documentation Page			Form Approved OMB No. 0704-0188	
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1. REPORT DATE DEC 2008	2. REPORT TYPE N/A	3. DATES COVERED -		
4. TITLE AND SUBTITLE Nanoscience for Inensitive Munitions Development			5a. CONTRACT NUMBER	
			5b. GRANT NUMBER	
			5c. PROGRAM ELEMENT NUMBER	
6. AUTHOR(S)			5d. PROJECT NUMBER	
			5e. TASK NUMBER	
			5f. WORK UNIT NUMBER	
7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES) Weapons and Materials Research Directorate Propulsion Science Branch Aberdeen Proving Ground, MD 21005			8. PERFORMING ORGANIZATION REPORT NUMBER	
9. SPONSORING/MONITORING AGENCY NAME(S) AND ADDRESS(ES)			10. SPONSOR/MONITOR'S ACRONYM(S)	
			11. SPONSOR/MONITOR'S REPORT NUMBER(S)	
12. DISTRIBUTION/AVAILABILITY STATEMENT Approved for public release, distribution unlimited				
13. SUPPLEMENTARY NOTES See also ADM002187. Proceedings of the Army Science Conference (26th) Held in Orlando, Florida on 1-4 December 2008, The original document contains color images.				
14. ABSTRACT				
15. SUBJECT TERMS				
16. SECURITY CLASSIFICATION OF:			17. LIMITATION OF ABSTRACT UU	18. NUMBER OF PAGES 10
a. REPORT unclassified	b. ABSTRACT unclassified	c. THIS PAGE unclassified		

Goals:

Develop predictive capability

Identify fundamental reaction mechanisms

Assess and advance methods and models



Approach:

Quantum Mechanics

Classical Molecular Dynamics

Empirical Methods

Benefit to the Warfighter:

Increased Lethality

Reduced vulnerability

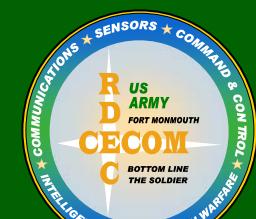
Faster, cheaper

IHEM01A/Theoretical Characterization of Energetic Materials

Develop enabling technologies to predict chemical and physical properties of EM to make a *priori* assessment of performance and vulnerability



US Army Research, Development and Engineering Command Environmental Quality Technology Ordnance Program “Design and Formulation of Novel Energetic Material Replacements for RDX”



Replace RDX with EM ingredients that have:

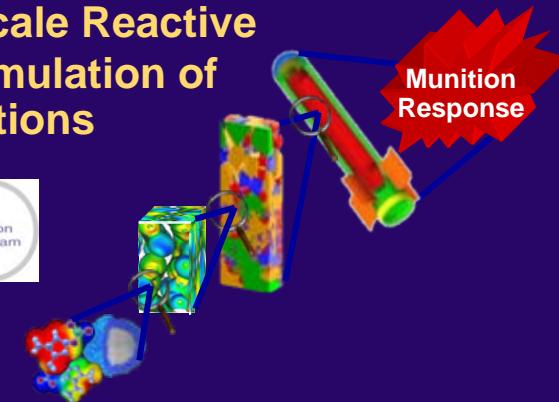
- Reduced toxicity, reduced hazard combustion/detonation products
- Equivalent performance properties
- Meet and/or exceed JROC 113-04 Insensitive Munition (IM) compliance requirements.

Office of Naval Research “Energetic Materials Modeling & Simulation Tools”



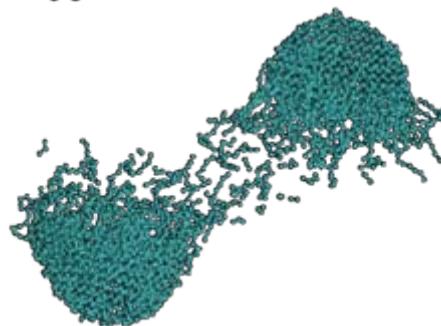
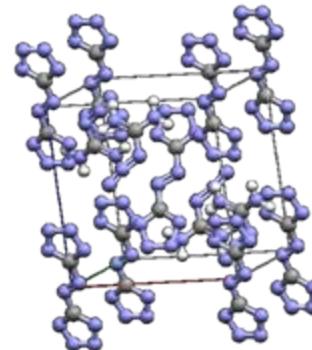
Assess and evaluate emerging models and software for EM research

HSAI for Multi-Scale Reactive Modeling and Simulation of Insensitive Munitions

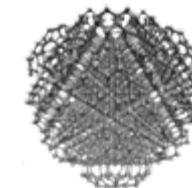
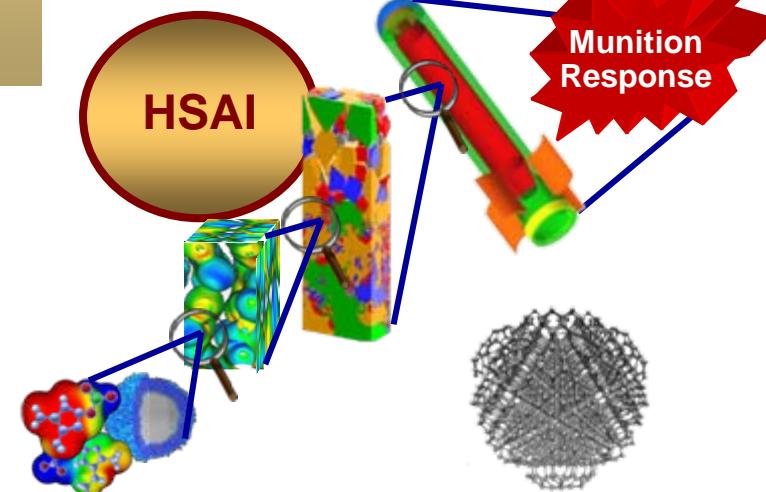
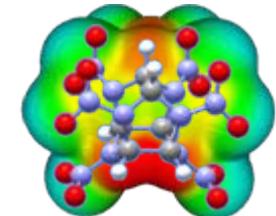
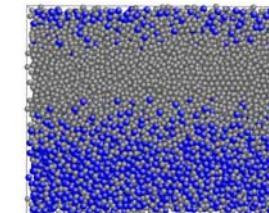
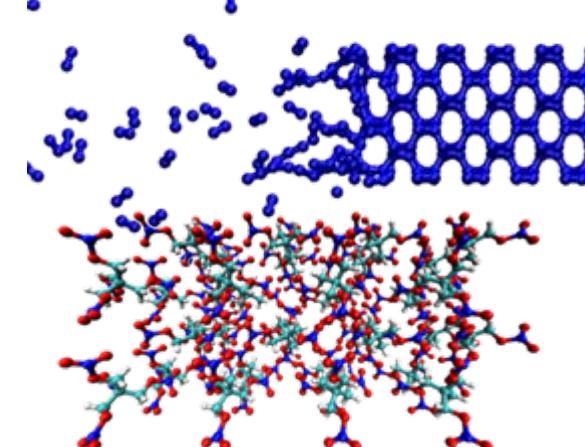




Collaborators and contributors



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Scott Weingarten
Anthony Yau*



Ludwig
Maximilians
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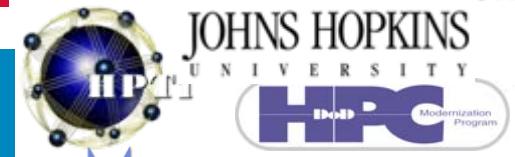
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UNIVERSITY
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INDIAN HEAD

Surface Warfare Center Division



NATIONAL LABORATORY



LAWRENCE LIVERMORE NATIONAL LABORATORY
Science in the National Interest



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Simulations used to explore dynamical behavior

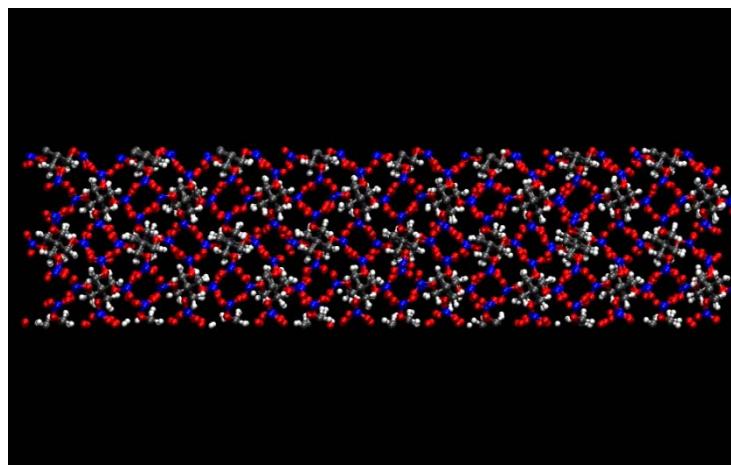
Classical MD



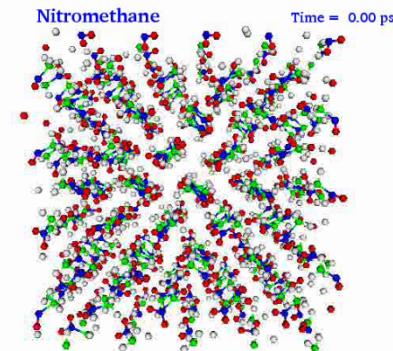
Self-sustained detonation of model explosive

Quantum MD

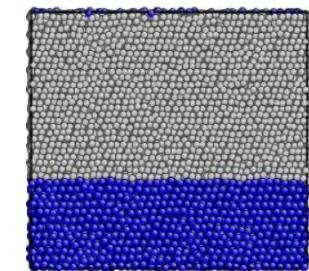
Shocked energetic materials



Conventional



Melting of nitromethane

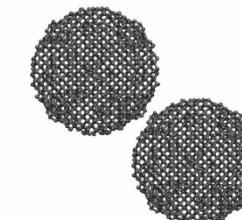


Thermal initiation of reactive material Ni/Al

Hypervelocity collisions of ND



Disruptive

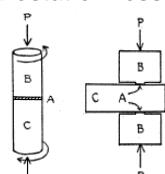


William D. Mattson, Radhakrishnan Balu, Betsy M. Rice, and Jennifer A. Ciezak

SBER: The release of excess energy stored in chemical bonds from structurally deformation by compressing, stretching or twisting the bond.

First Observed by
Bridgeman as Explosion of
Common Substances
Subjected to Pressure and
Shear

Effects of High Shearing Stress
Combined with Hydrostatic Pressure



Physical Review, 48 (1935) 825-47

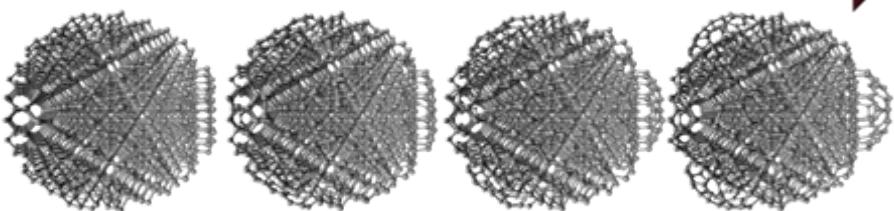
Former Soviet Union had Extensive SBER Program, with Particular Interest in Diamond

Structural Bond Energy Release in Energetic Materials as New Means for Designing Nonconventional High Explosives: An analysis of Soviet Research, Tech Report 1991. A. M. Al'tschuler, Technical Report TRC-91-0003, Technical Research Corporation, McLean, VA (1991).

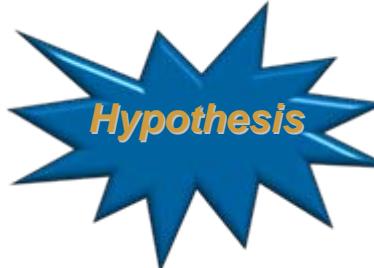
Potential Structural Bond Energy Materials

- ✓ Polymeric Nitrogen, an Extended Solid 8xTNT
- ✓ Nano-Diamonds, a Nanometer Scale Cluster 5xTNT

QM Diamond Surface Reconstruction with Buckyball Features



- Quantum simulations of smaller ND clusters show surface reconstruction to fullerene arrangement, core maintains diamond structure
- Calculations show tensile stress on the surface. Our calculations suggest core pressure in excess of 50GPa.



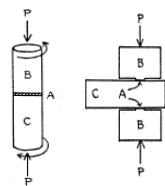
Sudden Disruption of ND Surface will Release Energy Stored in the Compressed Core

William D. Mattson, Radhakrishnan Balu, Betsy M. Rice, and Jennifer A. Ciezak

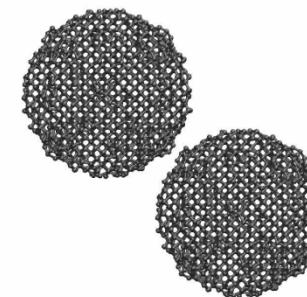
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Potential Structural Bond Energy Materials

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Testing the Hypothesis: Hypervelocity Collisions of ND

Perpendicular to Axis of Collision



0.08 ps



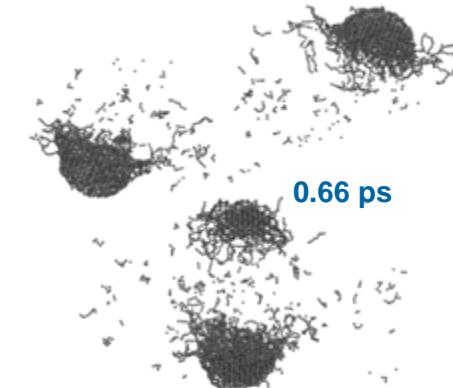
0.15 ps



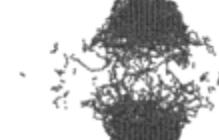
0.20 ps



0.40 ps

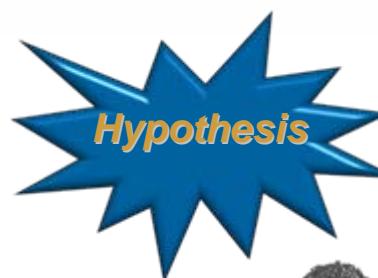


Along Axis of Collision



0.66 ps

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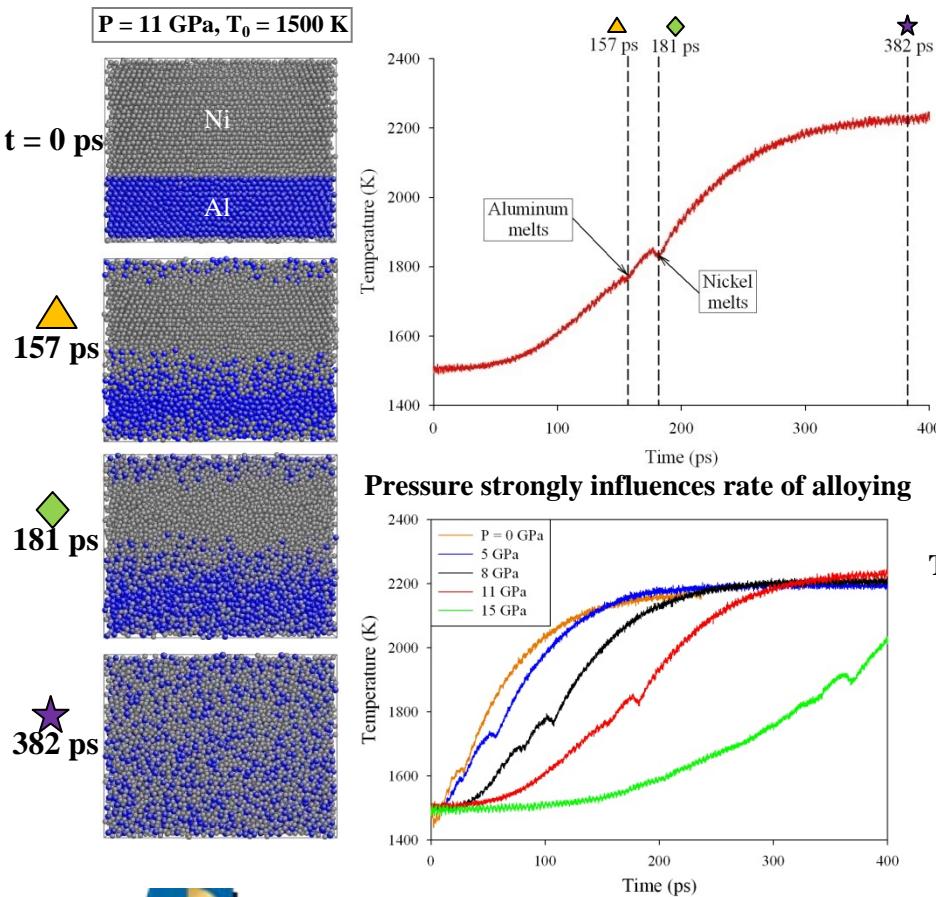


Sudden Disruption of ND Surface will Release Energy Stored in the Compressed Core

Identifying the Role of Pressure on the Response of Reactive Material to Thermal Initiation: A Molecular Dynamics Study

N. Scott Weingarten, William D. Mattson, Anthony D. Yau, and Betsy M. Rice (ARL) and Timothy P. Weihs, The Johns Hopkins University

Molecular dynamics (MD) simulations in the NPH ensemble
(number of particles, pressure and enthalpy held constant):



- Curves have same features: two “kinks” that correspond to melting of the two layers.
- Al melting always precedes Ni melting
- With increase in pressure, point of Al melting is less distinct

- Initial NVE-MD simulations at $T=1100$ K, reactions were quenched, in disagreement with experiment

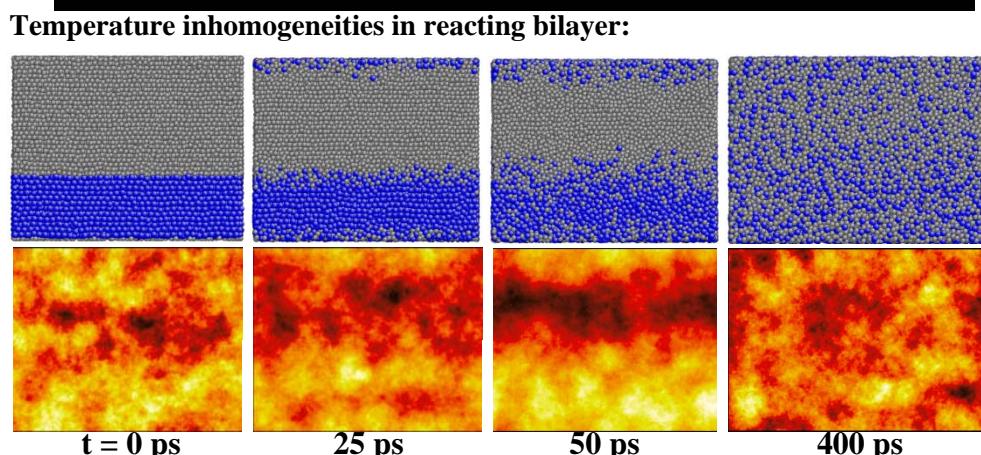
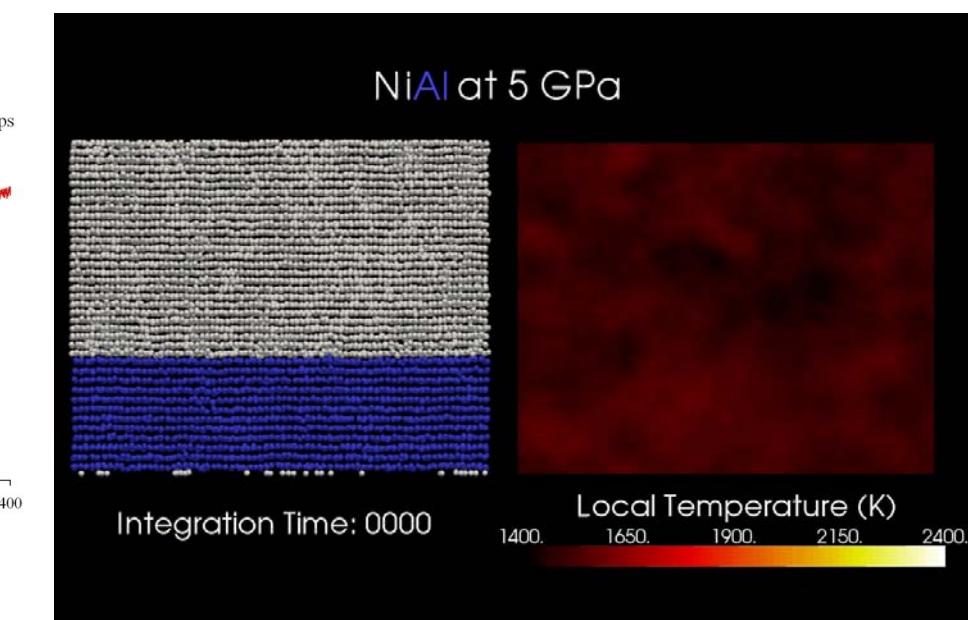
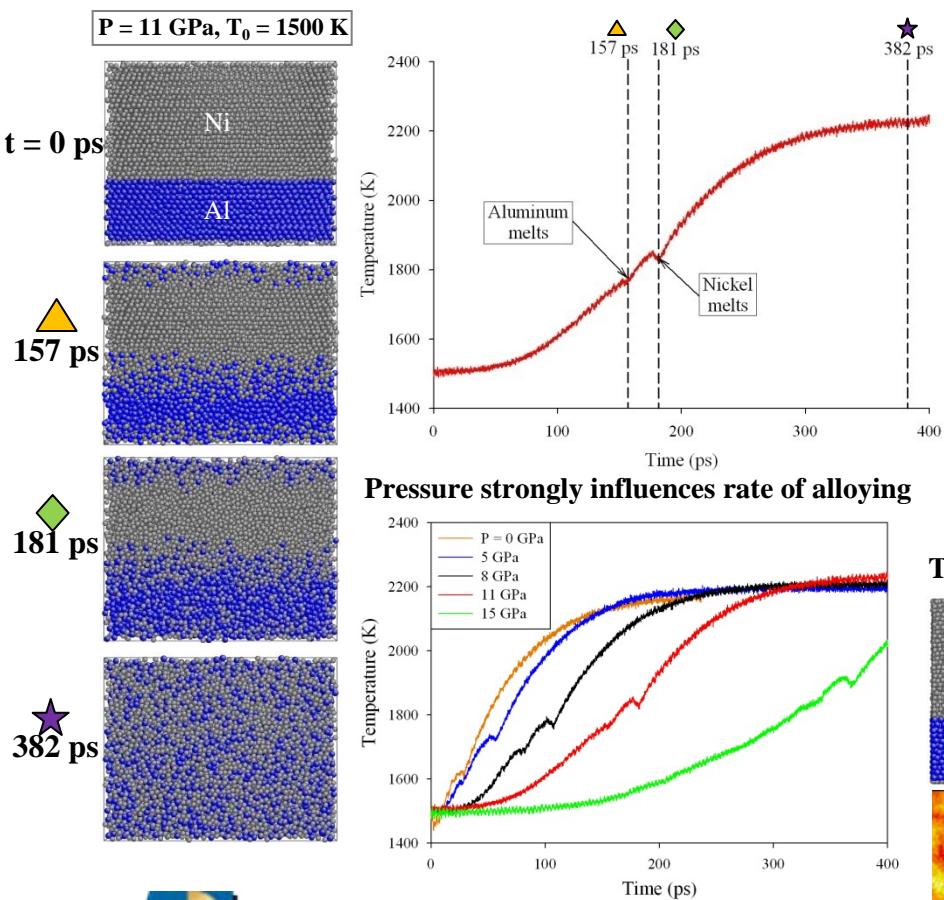
- Heating produced by reaction caused significant internal pressure which quenched the reactions

- NPH-MD simulations eliminate this problem: reaction rates are strongly influenced by imposed pressure.

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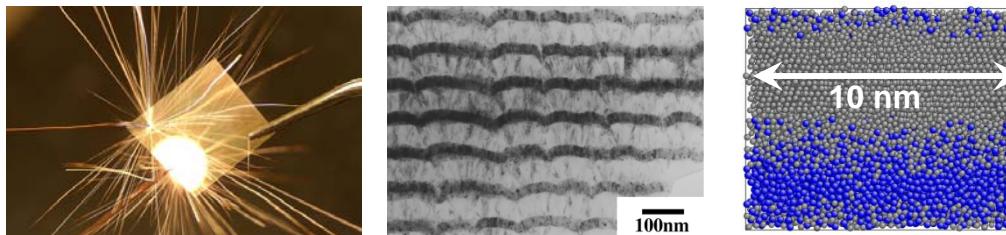
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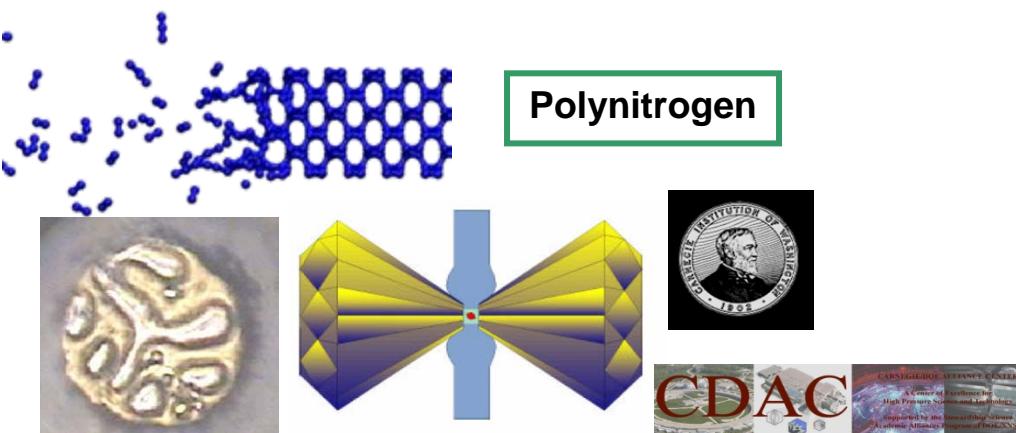
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Experimental Programs:

- Assist in experimental data interpretation
- Explore phenomena at inaccessible scales
- Validate or remedy models and methods



Nanolaminates of reactive materials



35 GPa

Questions?

Theoretical Programs:

- High Performance Computing Software
Application Institute for Multi-Scale
Reactive Modeling and Simulation of
Insensitive Munitions

